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Modern Chemistry 136 Chapter Test Name Class Date Chapter Test A, continued ____14. A pH meter measures the pH of a solution by measuring the a. amount of charge on the hydronium ion. b. color of the solution. c. concentration of the solution. d. voltage difference between two electrodes in the meter's probe. ____15. When you conduct an ...

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Modern Chemistry Chapter 15. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. mhardiell. Terms in this set (10) Self-ionization of water. two water molecules produce a hydronium and a hydroxide ion by transfer of a proton. pH.

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Modern Chemistry 141 Chapter Test Name Class Date Chapter Test B, continued 18. When a strong acid is titrated with a weak base, the pH of the solution at the equivalence point is than 7. 19. A is a highly purified solid used to check the concentration of a standard solution. 20. A 1 M solution of NaOH will have a pH that is

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Modern Chemistry Chapter 15 Test Review 25 multiple choice definition of self-ionization [H3O+] [OH-] = 1 x 10⁻¹⁴ definition and formula for calculating pH = -log[H3O+] pH >7 is a base; pH <7 is an acid; pH = 7 is neutral pH range is normally 0 to 14 calculate pH from [H3O+] calculate [H3O+] from pH definitions of indicators, transition interval, titration, neutralization, endpoint, standard solution acid-base titration monitors pH titration problems using VaMaega = VxMbqeb.

Modern Chemistry Chapter 15 Acid-Base Titration & pH

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Chapter 15 Test Acids And Bases Answers Chapter Tests Acids, Bases, and Solutions (continued) 28. Dip red litmus paper into each solution and watch for color changes. Then do the same with blue litmus paper. Any solution that turns red litmus blue is basic. Any solution that turns blue litmus red is acidic.

Chapter 15 Acids Bases Test Modern Chemistry

Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. ____ 7. A solution containing 35 g of Li 2SO 4 dissolved in 100 g of water is heated from 10°C to 90°C. According to information in the figure, this temperature change would result in a. an additional 5 g of Li 2SO 4 in ...

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Modern Chemistry 68 Chapter Test Name Class Date Chapter Test A, continued ____21. Metal X replaces the ions of metal Y from solution, but it cannot replace the ions of metal Z from solution. The order these metals should have in the activity series (from top to bottom) is a. x, y, z b. z, x, y c. y, x, z d. z, y, x ____22.

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Modern Chemistry 101 Chapter Test Name Class Date Chapter Test B, continued 18. The pressure exerted by each gas in a mixture is called the of that gas. 19. If the temperature and number of moles of a gas remain constant but the volume increases, the pressure of the gas will . 20. The lowest possible temperature, corresponding to zero on the ...