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1. all matter is composed of atoms. 2. atoms of a given element are identical. 3. atoms can't be subdivided, created, or destroyed. 4. atoms of different

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elements combine in whole-number ratios to form chemical compounds. 5. atoms are combined, separated, or rearranged in chemical reactions.  
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elements combine in simple,  
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3 SHORT ANSWER Answer the  
following questions in the space  
provided. 1. Explain the

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difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

3 Atoms: The Building Blocks of

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behavior of cathode rays in a glass tube containing gas at low pressure led scientists to conclude that the rays were composed of a. energy

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