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(Sections 5.1 - 5.4)

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*measurnig enthalpy*

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~~Thermodynamics~~

~~Chapter 4 Energy~~

~~Analysis of Closed~~

~~Systems~~ **The Laws of**

**Thermodynamics,**

**Entropy, and Gibbs**

**Free Energy**

~~Thermochemical~~

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oscillator (earthquake

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*Equilibria and Buffer*

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#### Thermochemistry 435

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Objectives • Apply

Hess's law of heat summation to find enthalpy changes for chemical and physical processes • Calculate enthalpy changes using standard heats of formation

Vocabulary • Hess's law of heat

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summation • standard  
heat of formation Key  
Equation •  $\sum H_f^\circ$   
(products)  $- \sum H_f^\circ$   
(reactants) Part  
A Completion

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THE FLOW OF  
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explains the

relationship between

energy and heat, and

distinguishes between

heat capacity and ...

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Section 1: The Flow  
of Energy – Heat and  
Work. What is

thermochemistry?

The study of the  
energy changes that



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occur in changes of state or during chemical reactions.

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Practice Problems

McLaughlin, Kimberly

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explains the  
relationship between  
energy and heat, and  
distinguishes between  
heat capacity and  
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